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#### Key indicators

Single-crystal X-ray study T = 291 KMean  $\sigma$ (C–C) = 0.006 Å R factor = 0.053 wR factor = 0.136 Data-to-parameter ratio = 15.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2006 International Union of Crystallography All rights reserved (3S,8S)-3,8-Diisopropyl-1,6-dioxa-4,9-diaza-5 $\lambda^5$ -phosphaspiro[4.4]nonane-2,7-dione

In the title compound,  $C_{10}H_{19}N_2O_4P$ , the molecule adopts a slightly distorted trigonal-bipyramidal structure. The molecules are linked by intermolecular  $N-H\cdots O$  hydrogen bonds, building a chain along the *b* axis parallel to the (001) plane.

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### Comment

The title compound, (I), was synthesized for studies of its spectroscopic characteristics and its biochemical activity. According to experimental evidence, this compound shows a 45% rate inhibition to tyrosinase (Yu, Liu, Fang, Zeng & Zhao, 2005), an important enzyme in the formation of melanin which is widely distributed in microorganisms, animals and plants (Seo *et al.*, 2003). This type of compound and its analogues may afford potent and efficient new inhibitors of tyrosinase.



The molecule adopts a slightly distorted trigonal-bipyramidal structure, with the two O atoms in axial positions and the N and H atoms in equatorial positions, as shown in Fig. 1. Major distortion from the ideal angle of  $120^{\circ}$  occurs for N1— P1—N2, which is  $124.34 (14)^{\circ}$ . This type of distortion is even more pronounced in the related compounds 6-dioxa-4,9-diaza-5-phosphaspiro[4.4]nonane (Meunier *et al.*, 1978), where the N—P—N angle is  $130.8^{\circ}$ , and (3S,8S)-3,8-dimethyl-1,6-dioxa-4,9-diaza- $5\lambda^{5}$ -phosphaspiro[4.4]nonane-2,7-dione (Yu, Liu, Fang & Zhao, 2005), where the N—P—N angle is  $127.6^{\circ}$ .

The crystal structure of (I) involves intermolecular N— H···O hydrogen bonds with an  $R_2^2(12)$  graph-set pattern (Etter *et al.*, 1990), leading to the formation of infinite chains along the *b* axis extending parallel to the (001) plane (Fig. 2 and Table 2).

# **Experimental**

Following a general procedure (Garrigues *et al.*, 1977), triethylamine (25.5 mmol) in tetrahydrofuran (5 ml) was added to a suspension of L-valine (1.00 g, 8.5 mmol) and phosphorus trichloride (1.2 g, 8.5 mmol) in tetrahydrofuran (20 ml) at 343 K over a period of 30 min. After stirring at 343 K for 2 h, the reaction mixture was



Figure 1

The molecular structure of compound (I). Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as spheres of arbitrary radii.

neutralized with NaHCO<sub>3</sub> to pH 7-8 and then extracted with ethyl acetate (4  $\times$  15 ml). The ethyl acetate layers were combined and dried with magnesium sulfate. The filtrate was concentrated at reduced pressure to give a white solid. The solid was dissolved in methanol and the solution was chromatographed (silica gel,  $3.0 \times$ 25 cm, chloroform-methanol-17% aqueous ammonia, 4:4:0.5 v/v/v). Recrystallization from acetone gave the desired compound, (I).

Z = 2

 $D_x = 1.273 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation

Block, colourless

 $0.24 \times 0.11 \times 0.09 \; \text{mm}$ 

4127 measured reflections

2471 independent reflections

2053 reflections with  $I > 2\sigma(I)$ 

where  $P = (F_0^2 + 2F_c^2)/3$ 

1069 Friedel pairs

 $\mu = 0.21 \text{ mm}^{-1}$ 

T = 291 (2) K

 $R_{\rm int} = 0.024$ 

 $\theta_{\rm max} = 25.5^{\circ}$ 

# Crystal data

 $C_{10}H_{19}N_2O_4P$  $M_r = 262.24$ Monoclinic, P21 a = 10.3925 (17) Å b = 6.0362 (10) Åc = 11.2115(19) Å  $\beta = 103.413(2)^{\circ}$ V = 684.1 (2) Å<sup>3</sup>

#### Data collection

Bruker APEX2 CCD area-detector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (SADABS; Bruker, 2004)  $T_{\min} = 0.953, T_{\max} = 0.981$ 

#### Refinement

Refinement on  $F^2$  $w = 1/[\sigma^2(F_o^2) + (0.0824P)^2]$ 
$$\begin{split} R[F^2 > 2\sigma(F^2)] &= 0.053 \\ wR(F^2) &= 0.136 \end{split}$$
 $(\Delta/\sigma)_{\rm max} < 0.001$  $\Delta \rho_{\rm max} = 0.77 \ {\rm e} \ {\rm \AA}^{-3}$ S = 1.03 $\Delta \rho_{\rm min} = -0.22$  e Å<sup>-3</sup> 2471 reflections 161 parameters Absolute structure: Flack (1983), H atoms treated by a mixture of independent and constrained Flack parameter: -0.05 (16) refinement

#### Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - \mathbf{H} \cdots A$
N2-H2···O4 <sup>i</sup>	0.86	2.27	2.989 (4)	141
$N1-H1\cdots O1^{ii}$	0.86	2.27	2.981 (4)	141

Symmetry codes: (i) x, y - 1, z; (ii) x, y + 1, z.



Figure 2

A partial packing diagram, showing the N-H...O hydrogen bonds (dashed lines) and the formation of infinite chains along the b axis extending parallel to the (001) plane. H atoms not involved in hydrogen bonds have been omitted for clarity. [Symmetry codes: (i) x, y - 1, z; (ii) x, y + 1, z].

H atoms attached to C and N atoms were positioned geometrically and treated as riding on their parent atoms, with C-H = 0.98(methine) or 0.96 Å (methyl), and N-H = 0.86 Å, and with  $U_{iso}(H) =$  $1.2U_{eq}(C,N)$  for methine H and NH, or  $1.5U_{eq}(C)$  for methyl H. The position of the H atom attached to the P atom was refined freely, with  $U_{iso}(H) = 1.2U_{eq}(P)$ . The absolute configuration of the compound was assigned by reference to the unchanging chiral centres C4 (S) and C7 (S) in the synthetic procedure and further confirmed by the refinement of the Flack (1983) parameter.

Data collection: SMART (Bruker, 2004); cell refinement: SAINT (Bruker, 2004); data reduction: SAINT (Bruker, 2004); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPIII (Burnett & Johnson, 1996), ORTEP-3 for Windows (Farrugia, 1997) and PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97.

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